

A Level Chemistry August 2011 Past Papers

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AS and A level Chemistry 2011 Past Papers - CIE Notes

Mark Scheme – General Certificate of Education (A-level) Chemistry – Unit 2: Chemistry in Action – January 2011 4 1(c)(i) (If any factor is changed which affects an equilibrium), the (position of) equilibrium will shift/move so as to oppose the change. OR (When a system / reaction in equilibrium is disturbed), the equilibrium

Final Mark Scheme - A Level Chemistry

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General Certificate of Education June 2011 CHEM5

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CHEM4 Unit 4: Kinetics, Equilibria and Organic Chemistry - June 2011 6 2(c)(ii) In pH values penalise fewer than 3 sig figs each time but allow more than 2 dp For values above 10, allow 3sfs - do not insist on 2 dp M1-XS mol KOH (= $(20 \times 10^{-3}) \times 0.154$) = 3.08×10^{-3} 1 If no subtraction: max 1 for correct use of volume No subtraction and no use of volume scores

A-level Chemistry Mark Scheme Unit 04 - Kinetics ...

water level is above dyes. Reject ref to reaction 1 (b) (i) C and D 1 (ii) insoluble Accept does not dissolve in water Reject ref to reaction Reject ref to not enough dye 1 (c) 52-55 67-68 0.76-0.82(1) Penalise use of cm once only in M1 + M2 Do not penalise more than 2sf in M1 - M3 Accept 1sf in M3 M3 CQ on M1 + M2, even for R f > 1 1 1 1

A-level OCR A Chemistry Past Papers - Past Papers

Mark Scheme - General Certificate of Education (A-level) Chemistry - Unit 5: Energetics, Redox and Inorganic Chemistry - June 2011 7 3(c)(i) $G = H - T S = -46 + 800 \frac{99.5}{1000} = 33.6$ or $33600 \text{ kJ mol}^{-1}$ -with J mol 1 1 1 1 mark is for putting in numbers with 1000 if factor of 1000 used incorrectly CE = 0

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Chem CHEM2 Spec 20) U : Che - A Level Chemistry

F325 Mark Scheme June 2011 1 Question Answer Mark Guidance 1 (a) (The enthalpy change that accompanies) the formation of one mole of a(n ionic) compound from its gaseous ions (under standard conditions) 2 IGNORE 'Energy needed' OR 'energy required' ALLOW as alternative for compound: lattice, crystal, substance, solid Note: 1st mark requires 1 mole

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Mark Scheme – General Certificate of Education (A-level) Chemistry – Unit 5: Energetics, Redox and Inorganic Chemistry – January 2011 3 Question Marking Guidance Mark Comments 1(a) Enthalpy change for the formation of 1 mol of gaseous atoms From the element (in its standard state) Enthalpy change to separate 1 mol of an ionic

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O level Chemistry 2011 Past Papers - CIE Notes

Mark Scheme for June 2011 Cherry Hill Tuition A Level Chemistry OCR (A) Paper 4. Mark Scheme . F321 Mark Scheme June 2011 Question Answer Mark Guidance 1 (a) particle rel charge rel mass position proton +1 1 nucleus neutron nil/0 1 nucleus

electrons -1 1/2000 in shells ...

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A Level Chemistry August 2011 Past Papers

Mark Scheme - General Certificate of Education (A-level) Chemistry - Unit 2: Chemistry In Action - June 2011 6 2(d) M1 The enthalpy change / heat change at constant pressure when 1 mol of a compound / substance / product M2 is formed from its (constituent) elements M3 with all reactants and products / all substances in standard states

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