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13. The SI base unit for mass is the 14. The SI derived unit for volume is the liter meter Chapter Assessment is a ratio that compares the mass of an object to its volume. include all known digits plus one estimated digit. Chemistry. Matter and Change Chapter 2

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As a scuba diver ascends, pressure decreases. A decrease in pressure results in an increase in volume. If a diver holds his or her breath

Chemistry Matter And Change Chapter 15

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Chapter Summaries - Chemistry Matter and Change

SPayette. Chemistry Matter and Change: Chapter 13. kinetic-molecular theory. elastic collision. temperature. diffusion. explains the properties of gases in terms of the energy, size,.... a collision in which kinetic energy may be transferred between.... a measure of the average kinetic energy of the particles in a....

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Ch 13 - States of Matter. 13.1 Gases. Kinetic-molecular theory, particle size, elastic collision, temperature, diffusion, Graham's Law of Effusion (smaller molecules of gas diffuse faster) pressure,

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barometer, pascal (Pa, kPa), atmosphere (atm), torr = mmHg, Dalton 's Law of Partial Pressures

Chapter 13 Solutions | Chemistry: The Molecular Nature Of ...

13.1 Section Check. •Relate number of particles and volume using Avogadro's principle. mole: an SI base unit used to measure the amount of a substance; the amount of a pure substance that contains 6.02×10^{23} representative particles •Relate the amount of gas present to its pressure, temperature, and volume using the ideal gas law.

How much carbon dioxide gas, in grams, is in a 1.0-L ...

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