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PhET: Concentration Watch your solution change color as you mix chemicals with water. Then check molarity with the concentration meter. What are all the ways

you can chan...

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Concentration and Molarity PhET Labs. Name: _____ Part 4: Calculating Molarity . Using the simulation and the formula for Molarity on the front, complete the table below. ... Concentration and Molarity Post-Lab Exercises . 1. Adding pure water to a saturated solution (with no solids) would cause the concentration of that solution to . increase /

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Describe the relationships between volume and amount of solute to concentration. Explain how solution color and concentration are related. Calculate the concentration of solutions in units of molarity (mol/L). Use molarity to calculate the dilution of solutions. Compare solubility limits between solutes. Version 1.2.10

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Molarity This is an introduction that uses the molarity Phet to illustrate how concentration changes PhET Molarity: Solutions and Concentration HChem304B MNVA

Concentration and Molarity PhET Labs

Concentration and Molarity PhET-Chemistry Labs. Introduction: How is rock candy made? Explain. Some handy vocabulary for you to define: Give an example from the sim lab for starred words *Solute: the substance that is _____ into a solvent.

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Watch your solution change color as you mix chemicals with water. Then check molarity with the concentration meter. What are all the ways you can change the concentration of your solution? Switch solutes to compare different chemicals and

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find out how concentrated you can go before you hit saturation!

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Substituting the given values for the terms on the right side of this equation yields:
 $C_2 = 0.850 \text{ L} \times 5.00 \text{ mol L}^{-1} / 1.80 \text{ L} = 2.36 \text{ M}$
 $C_2 = 0.850 \text{ L} \times 5.00 \text{ mol L}^{-1} / 1.80 \text{ L} = 2.36 \text{ M}$. This result compares well to our ballpark estimate (it's a bit less than one-half the stock concentration, 5 M). Check Your Learning.

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C5.4.3 explain how the concentration of a solution in mol/dm^3 is related to the mass of the solute and the volume of the solution and calculate the molar concentration using the formula: $\text{concentration (mol/dm}^3\text{)} = (\text{number of moles in solute}) / (\text{volume (d...})$

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Is a concentration term for solution is the number of moles of solute dissolved in one liter of solution. To calculate the molarity of a solution, divide the moles of solute by the volume of the solution.

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