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the thermal energy of a system is equal to the sum of the work done on it and the heat added to it; this is a statement of the \_\_\_\_\_ ... physics study guide (2013-14 orgeron) chapter 22 evaluation; chapter 15 evaluation;

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Read Free Chapter 12 Study Guide Thermal Energy the heat that is added to the object, minus the work done by the object heat engine a device that continuously converts thermal energy to mechanical energy; requires a high-temperature thermal energy source, a low-temperature receptacle (a sink), and a way to convert the thermal energy into work

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Heat. -the transfer of thermal energy from one piece of matter to another. Conduction. - the transfer of thermal energy between things that are touching. (particles of matter bumping into each other) Convection. -the transfer of thermal energy by the movement of liquids or gases.

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It will boil, thanks to thermal energy. This quiz/worksheet combo will help you assess your understanding of the properties of thermal energy. ...

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quantity of energy needed to change a unit mass of a substance from a liquid to a gaseous state at the boiling point first law of thermodynamics change in internal or thermal energy is equal to the heat added and work done on system. same as law of conservation of energy

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states that the change in thermal energy of an object is equal to the heat that is added to the object, minus the work done by the object heat engine a device that continuously converts thermal energy to mechanical energy; requires a high-temperature thermal energy source, a low-temperature receptacle (a sink), and a way to convert the thermal energy into work Chapter 12: Thermal Energy - Physics with Richard at ...

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## **What is Thermal Energy? - Study.com**

Thermal energy is energy possessed by an object or system due to the movement of particles within the object or the system. Thermal energy is one of various types of energy, where ' energy ' can be...

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Temperature and Thermal Energy. STUDY. PLAY.  
Thermal equilibrium. The state at which the rate of energy flow between two objects is equal and the objects are at the same temperature. Heat.

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the thermal transfer of energy by electromagnetic waves through the vacuum of space specific heat the amount of energy that must be added to a material to raise the temperature of a unit mass by one temperature unit; is measured in  $J/[(kg)(K)]$

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### **Thermal Energy Ch 12 Study Guide - Skinny Ms.**

The thermal energy of the hotter object is decreased, and the thermal energy of the cooler object is increased. Energy always flows from the hotter object to the cooler object. Energy never flows from a colder object to a hotter object. Heat is the energy that flows between two objects as a result of a difference in temperature.

## Quiz & Worksheet - Properties of Thermal Energy | Study.com

the change in thermal energy of an object is equal to heat that is added to an object minus the work done by the object.  $\Delta U = Q - W$ . first law of thermodynamics. device that is able to continuously convert thermal energy to mechanical energy. heat engine. measure of disorder in a system. (change in entropy:  $\Delta S = Q/T$ )

### Physics Ch 12: Thermal Energy, Section 1: Temperature and ...

Thermal energy refers to the TOTAL energy of all of the particles in a substance. Therefore, the thermal energy of a substance depends on the amount, while the temperature does not. Heat is the transfer of thermal energy from one object/substance to another, which results in a temperature change. Click again to see term

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States of Matter and Thermal Energy Study Guide \_\_\_  
LT 1: I can identify, define, give examples, and illustrate the states of matter of substances. \_\_\_ LT 2: I can predict the effect of adding or removing thermal energy. \_\_\_ LT 3: I can explain the transfer of thermal energy between particles through conduction.

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